

EE122B

Introduction to Biomedical Electronics

Miscellaneous Biomedical Sensors and Interfaces

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Content

Selected biomedical sensor applications, their transduction mechanisms, and associated interface circuitry.

- Temperature sensing
- Displacement and motion sensing
- Electrochemical sensors
 - Potentiometric
 - Amperometric

Technology

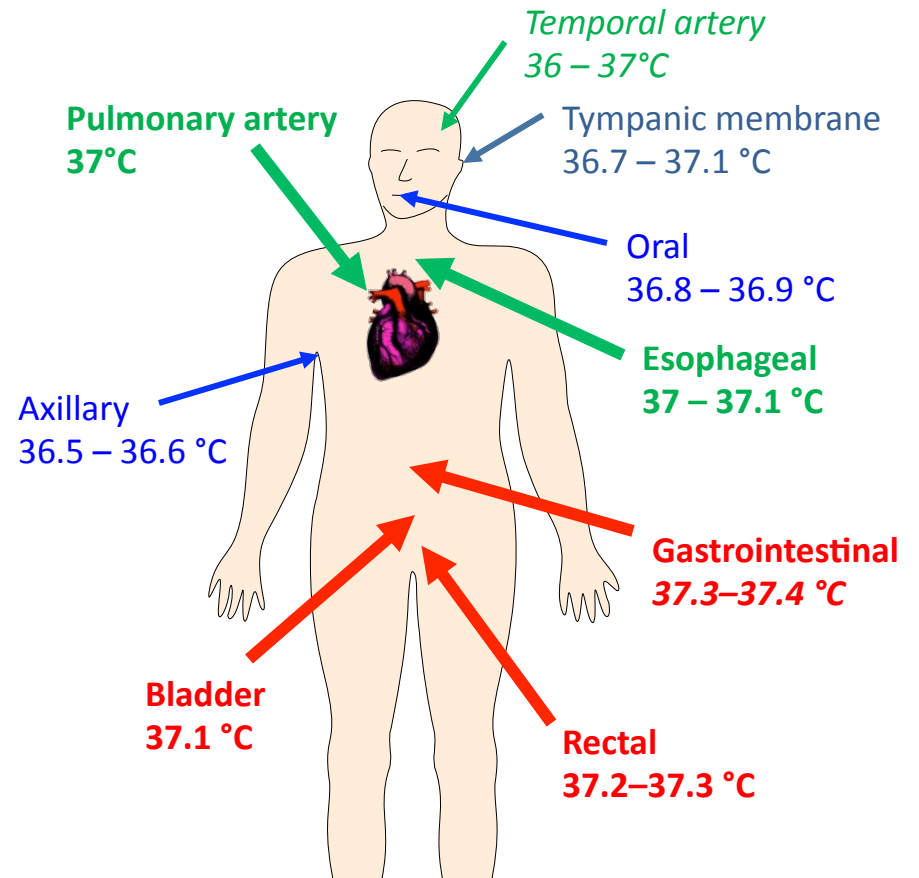
Temperature Sensing

Temperature Physiology

- We (mammals) are a warm-blooded species (*endotherm*).
- Thermoregulation is critical for biological processes to function properly (e.g., protein function).
Temperature is affected by many processes, some of which are of high diagnostic value (fever, local inflammation,...).
- Thermoregulation is mainly under the control of *hypothalamus*.
Temperature-sensitive neurons in the hypothalamus, skin and spinal cord relay temperature to the hypothalamus.
- At rest, the body generates ≈ 90 W of heat, and maintain an average temperature of 37°C .
- Heat transfer mechanisms are *radiation*, *conduction* and *convection* and *evaporation of perspiration*.
- *Cooling* is mainly achieved through *perspiration*, aided by vasodilation.
- *Heating* is performed through increased metabolism, muscle activity (shivering), vasoconstriction and shut down of perspiration.

Measurement Locations

- Core temperature
 - Catheter (pulmonary artery)
- Pseudo-core
 - Rectal
 - Gastrointestinal (pill)
 - Esophageal
 - Oral (sublingual)
 - Temporal artery
 - Tympanic membrane
- Surface temperature
 - Skin temperature



Desired measurement range: 30-44°C
Resolution: 0.1°C
Bandwidth: DC - 0.1Hz (up to 1Hz)

Derived from : Mary Frances D. Pate, *Thermal Regulation* in 'Principles and Practice of Pediatrics', 1999; Suleman, *Anesthesia & analgesia*, 95, 2002; Byrne, *Br. J. Sports Med*, 41, 2007.

Measurement Technology

- Resistance-based
 - Resistor Temperature Devices (RTD)
 - Thermistors
- Thermocouples
- Radiation-based
- Fiber Optic-based

Modulation of total internal reflection, fluorescence or absorption in semiconductor.
- Thermochromic

Temperature-based color change (liquid crystals, thermometer patches).
- Many other technique sensitive to temperature...

Thermistors & RTDs

- Resistance-based temperature sensors
- Resistor Temperature Device (RTD)
 - Positive temperature coefficient of resistance (TCR, or TEMPCO).
 - Fairly linear, very accurate.

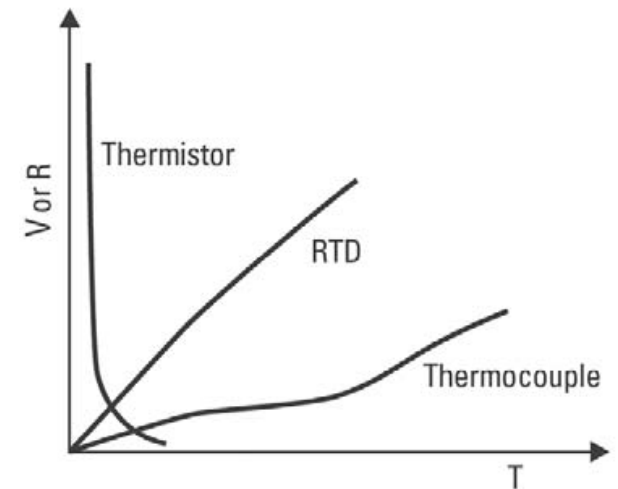
Example platinum RTD : TCR = $0.00385 \Omega/\Omega/^{\circ}\text{C}$.

For 100Ω (typical) at 0°C , **0.38Ω per $^{\circ}\text{C}$**



- Thermistor
 - Large negative temperature coefficient.
 - Good sensitivity, but not linear.

$$R = R_0 e^{\beta(1/T - 1/T_0)}$$



Source: Agilent Application Note 290

Example YSI 401 (ceramic): $R_0 = 2.25 \text{ k}\Omega$, $\beta = 3976 \Omega$ (at 25°C) $\rightarrow$ **-100Ω per $^{\circ}\text{C}$**

- Typically used for catheters, rectal, esophageal probes (example: series 400 from YSI)

Thermistor Conditioning

- Problem: “Self-heating”

Current through thermistor should stay low to avoid self-heating.

Bad: inaccurate temperature measurement

Worse: destruction of sensor

(thermal run-off if NTC)

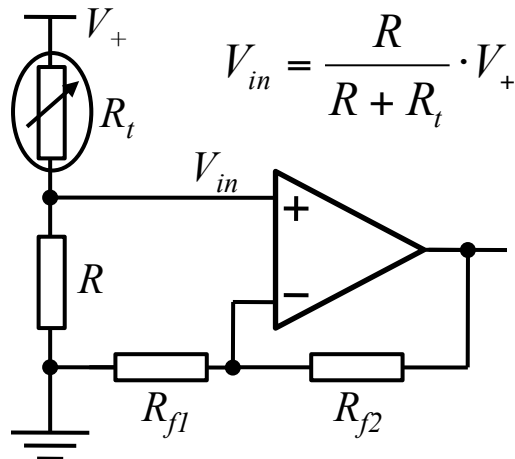
- Linearization

- Digital, look-up table (third-order polynomial, Steinhart-Hart equation)

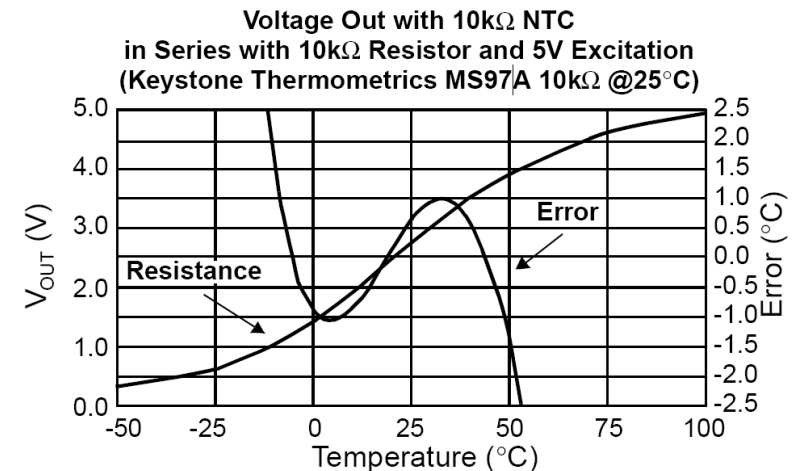
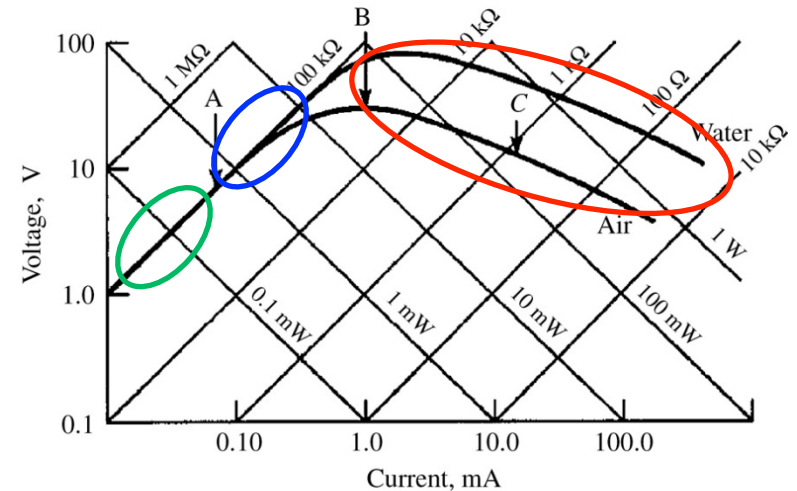
- Analog

http://en.wikipedia.org/wiki/Steinhart-Hart_equation

Example: series resistance linearizes behavior around R .



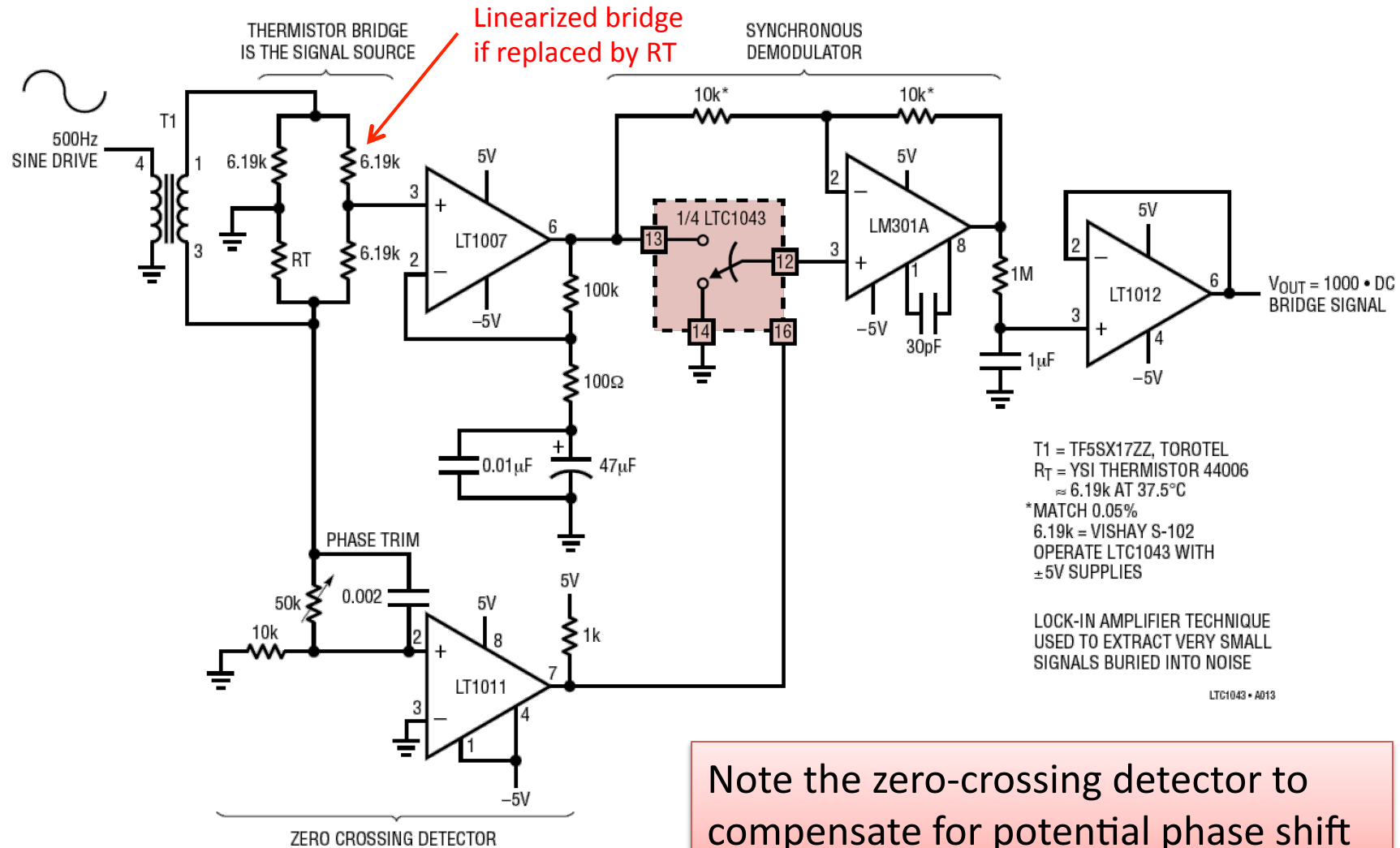
Maximum linearization around T for $R = R_t(T)$



Source: Microchip AN685

Thermistor Conditioning

Synchronous demodulator based on switched-cap block (Linear Technologies LT1043)



Note the zero-crossing detector to compensate for potential phase shift

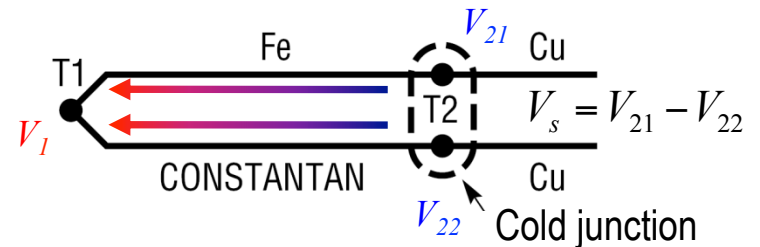
Source: LT1043 datasheet

Thermocouples

- An electromotive force (EMF) appears at the junction of two dissimilar metals. This voltage is a function of the temperature at the metal junction, and at the other end of the metal conductors.

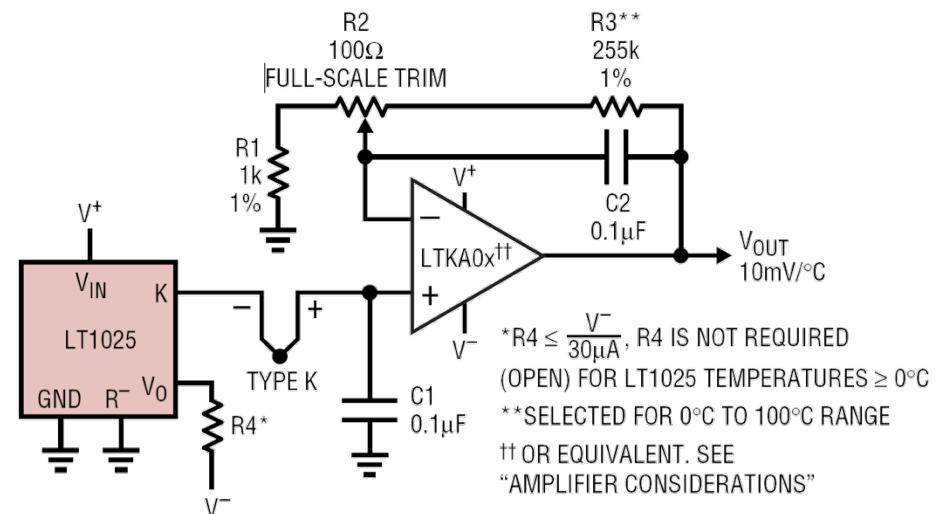
$$V_s = \alpha(T_1 - T_2)$$

α : Seebeck coefficient



Accurate measurements require the “cold junction” to be at a *reference* (ice) or a *known temperature*.

- Typical sensitivities:
 Type K (chromel-alumel): $\alpha = 41 \mu\text{V}/^\circ\text{C}$
 Type J (iron-constantan) : $\alpha = 50 \mu\text{V}/^\circ\text{C}$
- Small size, fast response time
 10 - 20mm diameter, 1 millisecond
- Special IC provide cold-junction compensation (LT1025, AD597)



Source: LT1025 datasheet

Zero-Drift Amplifiers

- DC performances of conventional op-amps are usually not stellar (offsets, drift, $1/f$ noise).
- Precision, chopper amplifiers are optimized for low frequency operation.
- Two main approaches:
 - **Auto-nulling**: periodic sampling of input offset and compensation.
 - **Chopper** amplification: input modulation, AC amplification, demodulation

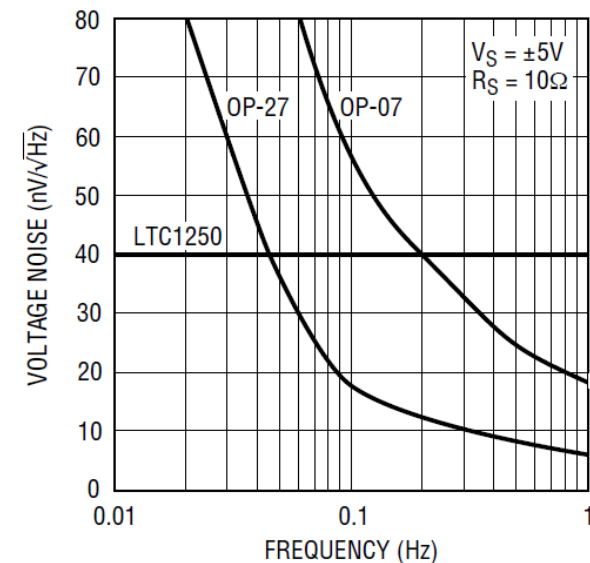
Other circuits: MAX4238, LMV2011, OPA734,...

Example: LT1250

Offset voltage : $5\ \mu\text{V}$

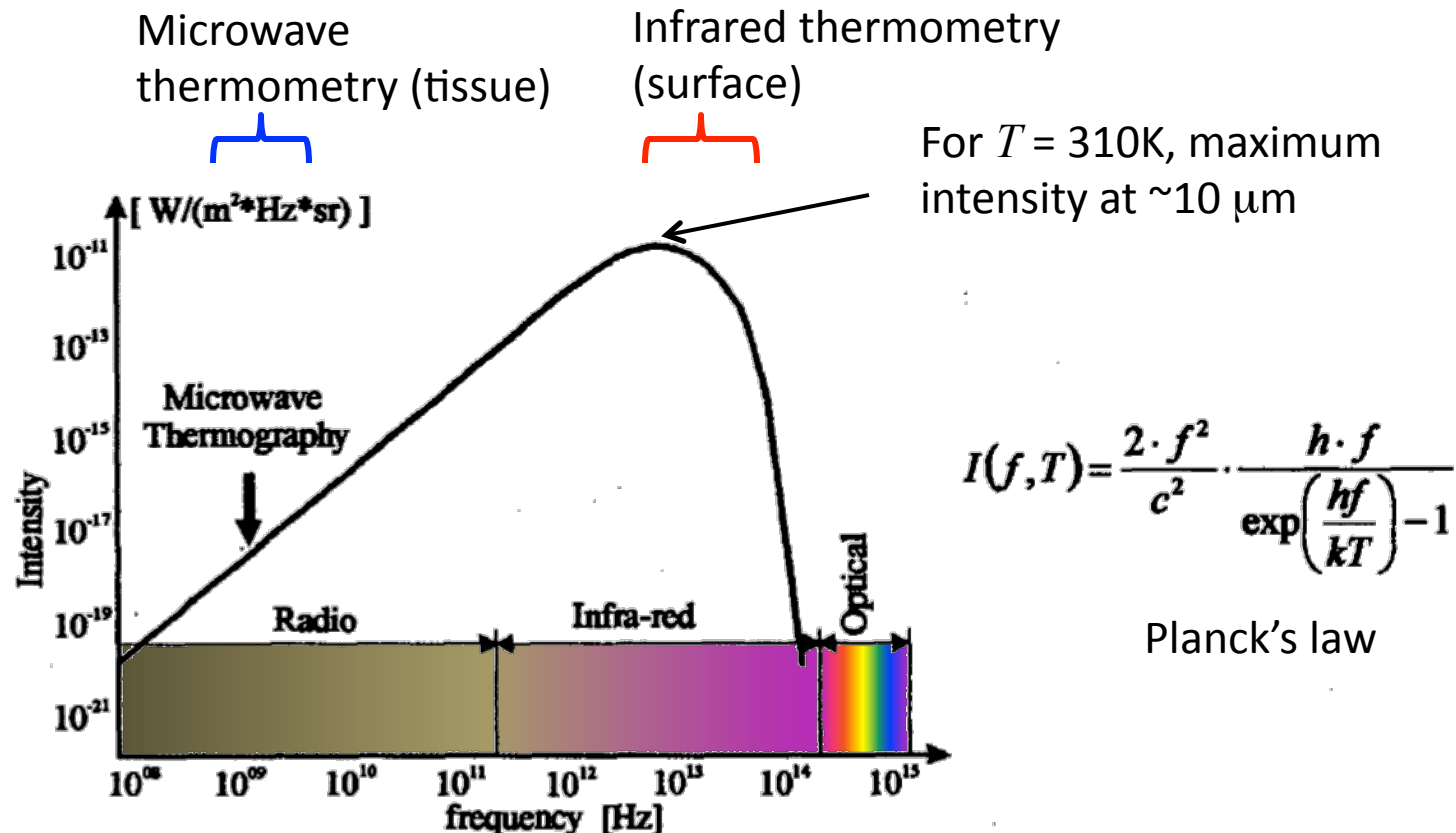
(LT1056 offset voltage : $150\ \mu\text{V}$)

Noise:



Radiative Measurements

Based on radiation emitted by any body at temperature above absolute zero (AKA black body radiation).



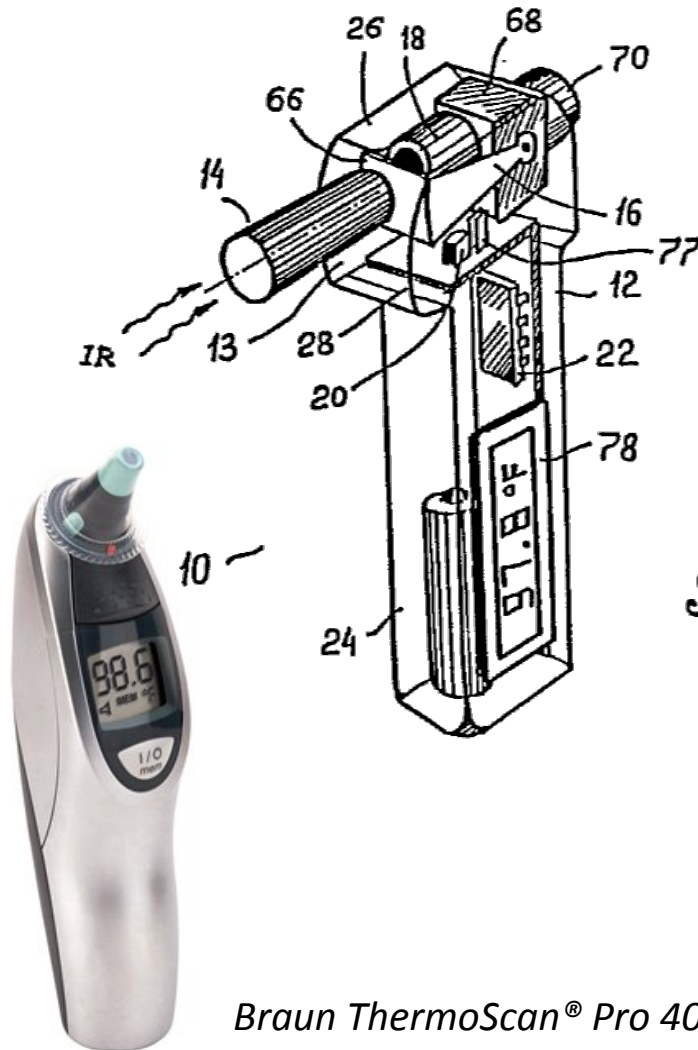
B. Stec, W. Susek, A 4.4 GHz microwave thermometer with compensation of reflection coefficient, 13th IEEE International Conference on Microwaves, Radar and Wireless Communications, 2000.

Infrared Thermometry

- Infrared sensors (quantum or thermal) measure surface temperature.
- Tympanic membranes and temporal regions are usually used.
They are perfused with blood coming directly from the heart, which also perfuses the hypothalamus.
- Sensitive to surface emissivity ε .
5% change equals to 1°C at 300K and 3 μm
Skin emissivity is generally 0.92 - 0.97
- Also used in imaging
Single and multispectral

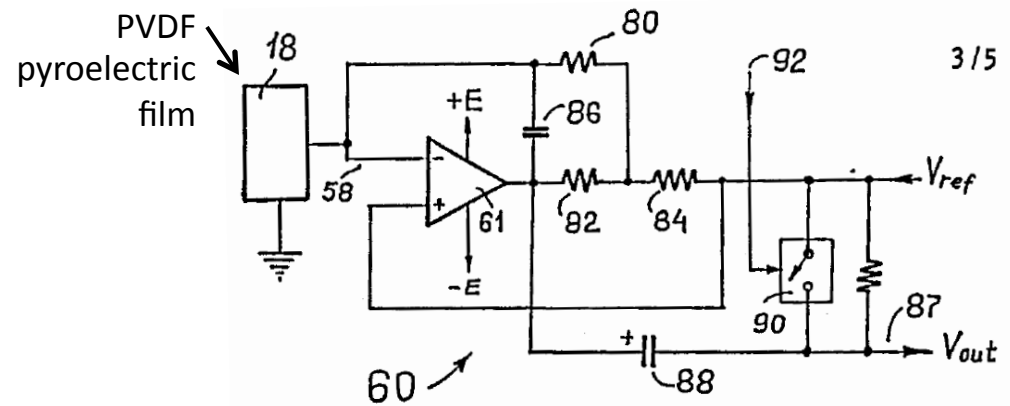
Infrared Thermometer

US Patent 4,797,840 (1989, ThermoScan, bought by Braun)

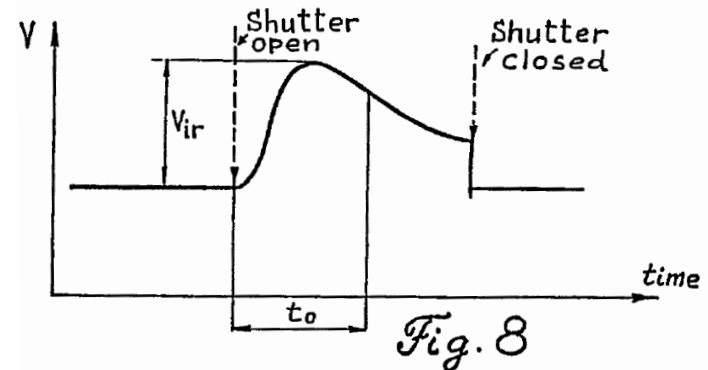


Braun ThermoScan® Pro 4000

Sensing circuit (transimpedance amplifier)



Typical sensor response



Mechanical sensing

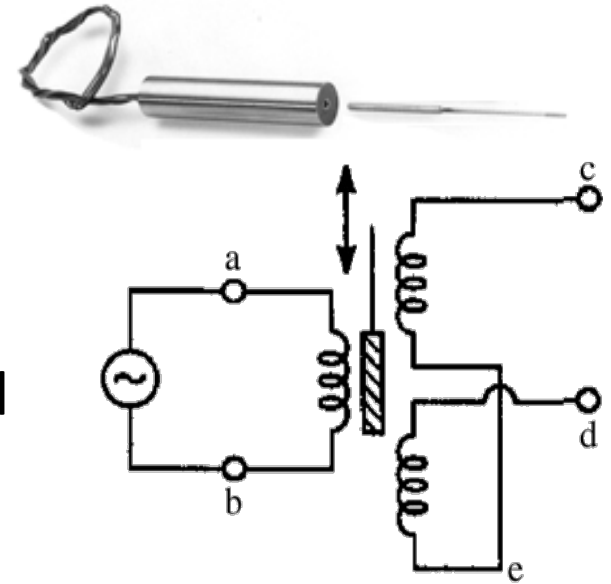
Displacement and Motion Sensing

Displacement and Motion Sensors

- Displacement and motion sensor are at the core of most mechanical sensing application.
Example: Pressure can displace a membrane
- Displacement, velocity, acceleration and force are all interrelated ($a = \dot{v} = \ddot{x}, F = ma$).
- Wide applicability to biomechanics, rehabilitation and physiologic monitoring,...
- We will review three useful sensors:
 - Linear Variable Differential Transformer (LVDT, displacement)
 - Strain gauges
 - Accelerometers

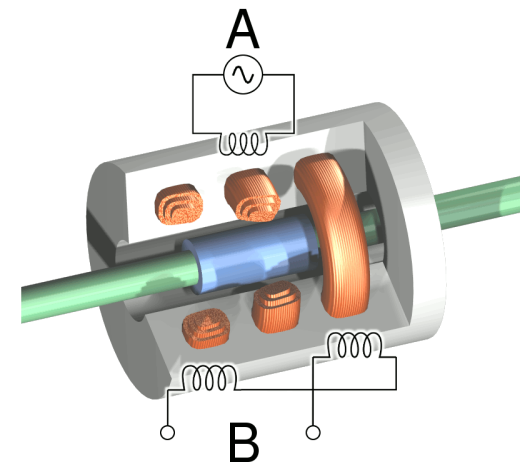
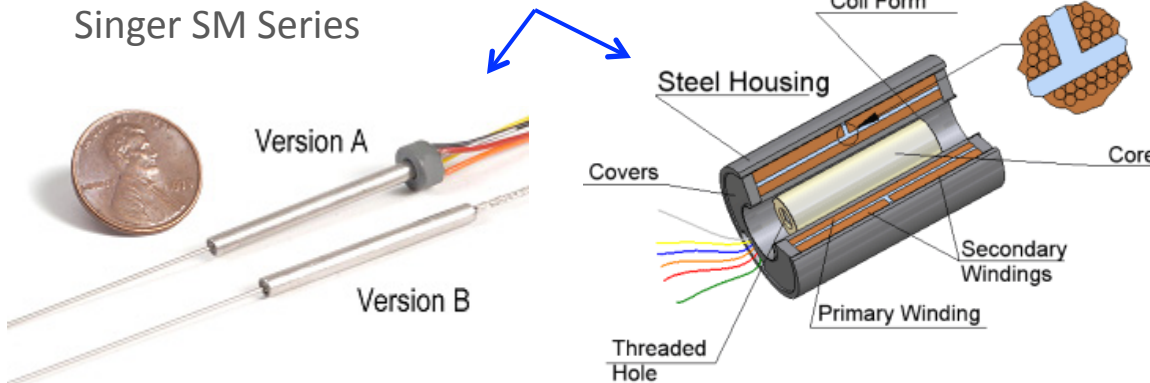
Displacement Sensor - LVDT

- An electromechanical sensor for displacement is the *Linear Variable Differential Transformer* (LVDT).
- A ferromagnetic core moves between two transformers, splitting the excitation signal proportionally.
- Wide range of spans (millimeter to meter) and resolutions (down to μm).



Note: even at full scale, the core still overlaps the three coils (linear range).

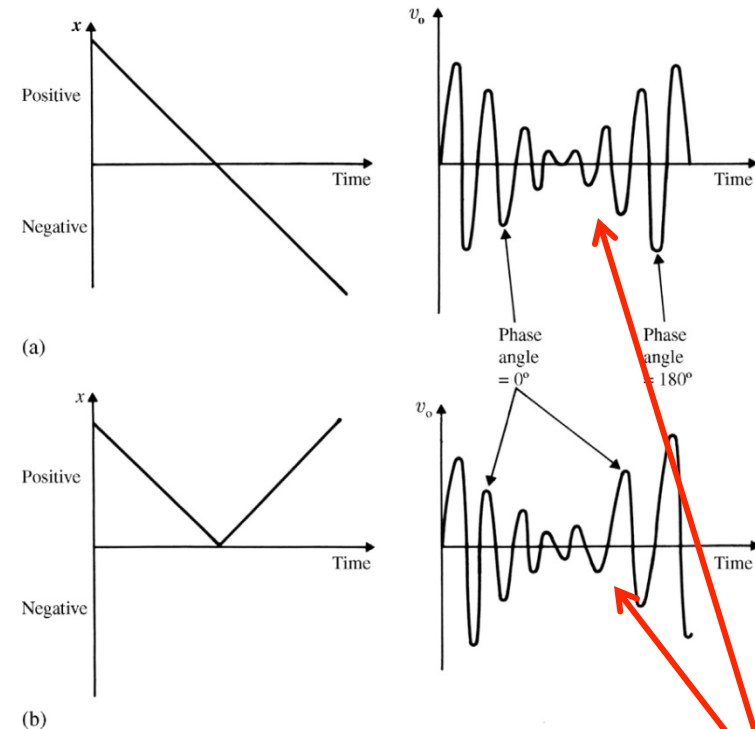
Miniature LVDT ($\varnothing 2.3 \text{ mm}$,
Singer SM Series)



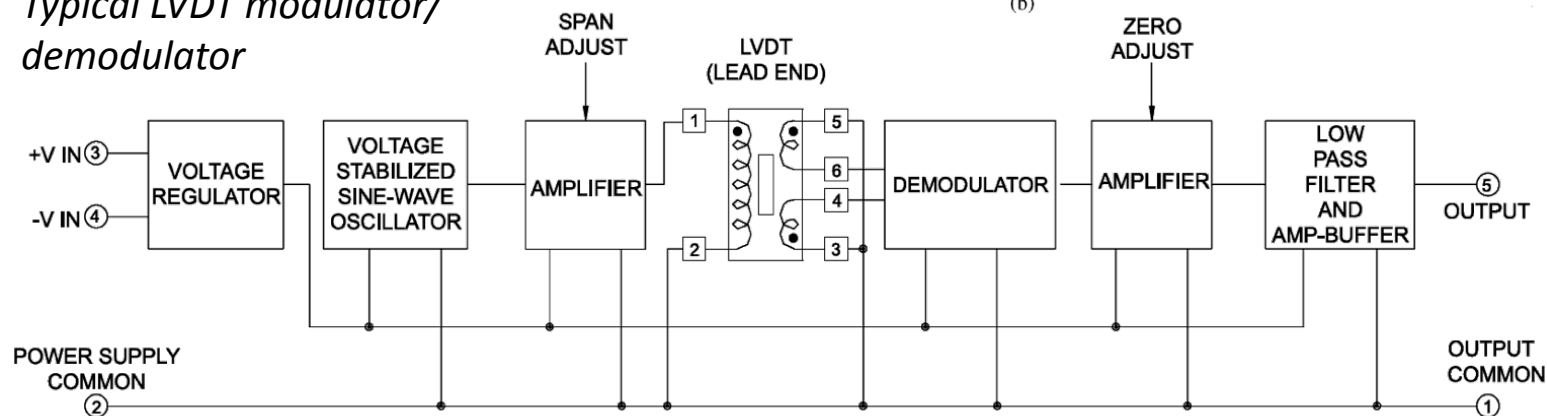
Sources: <http://www.transtekinc.com>, <http://www.singer-instruments.com>, Wikipedia

LVDT – Signal Conditioning

- LVDTs require a phase-sensitive demodulation to differentiate sides.
Simple diode rectifiers blind to phase-inversion at neutral point.
- Typical excitations are in the 50 Hz – 20 kHz range, at levels of 1 - 24V.



Typical LVDT modulator/demodulator



A simple rectifier would give the same output

Single-chip solution to LVDT modulation/demodulation: **ADI 598/698**

Strain Gauges

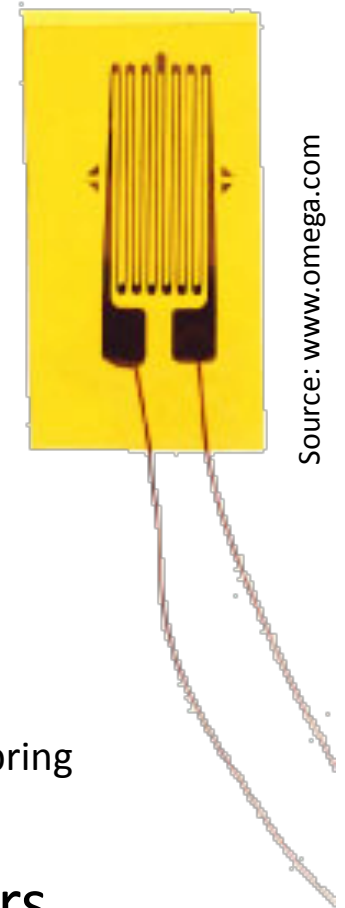
- Strain gauges are resistance sensitive to strain (relative length change).
- Defined by their *Gauge Factor* (GF):

$$GF = \frac{\text{relative resistance change}}{\text{relative length change}} = \underbrace{(1 + 2\mu)}_{\text{Geometrical effect}} + \underbrace{\frac{d\rho/\rho}{\varepsilon}}_{\text{Piezoresistive effect}}$$

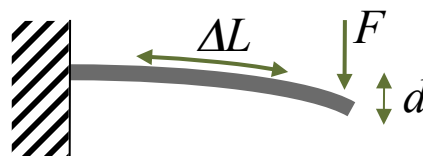
μ : Poisson ratio
 ρ : resistivity
 ε : strain (dL/L)

Typical gauge factors:

Metal: ~ 2
 Semiconductor: 80-200



- Through deformation of a beam, they can measure force:



$$\Delta R \propto \Delta L \propto d \propto kF \quad \text{where } k \text{ is the spring constant}$$

- Typical application: weighing scales, pressure sensors, biomechanical measurements (e.g., heel strike force).

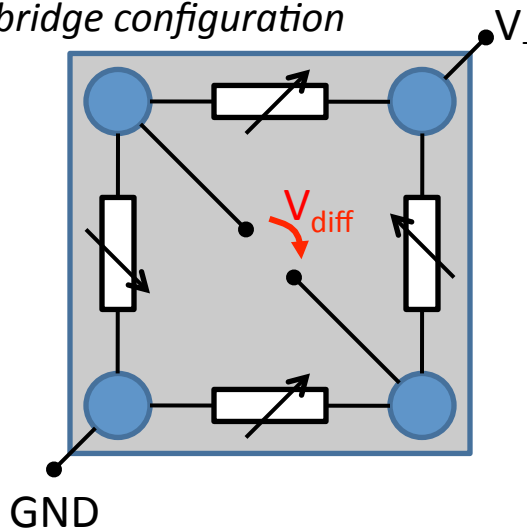
Application: Scales and Force Plates

- Scales measure static body weight.
- Force plates add dynamic capabilities and measurement of center of pressure and independent component of force (F_x , F_y , F_z , M_x , M_y , M_z).
- Used in rehabilitation, gait and balance measurements.

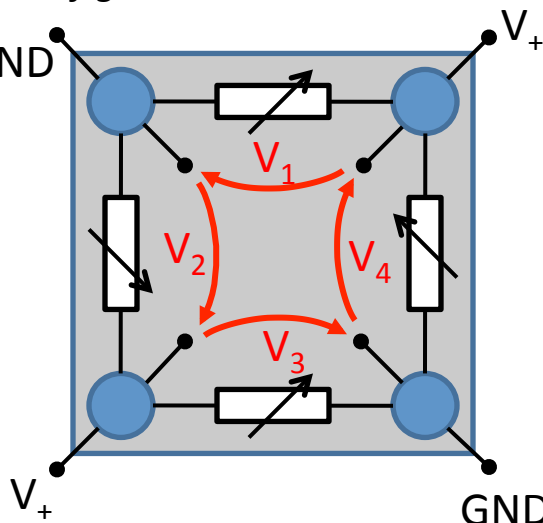


Nintendo Wii Fit™

*Weight measurement
bridge configuration*



*Center of pressure measurement
configuration*

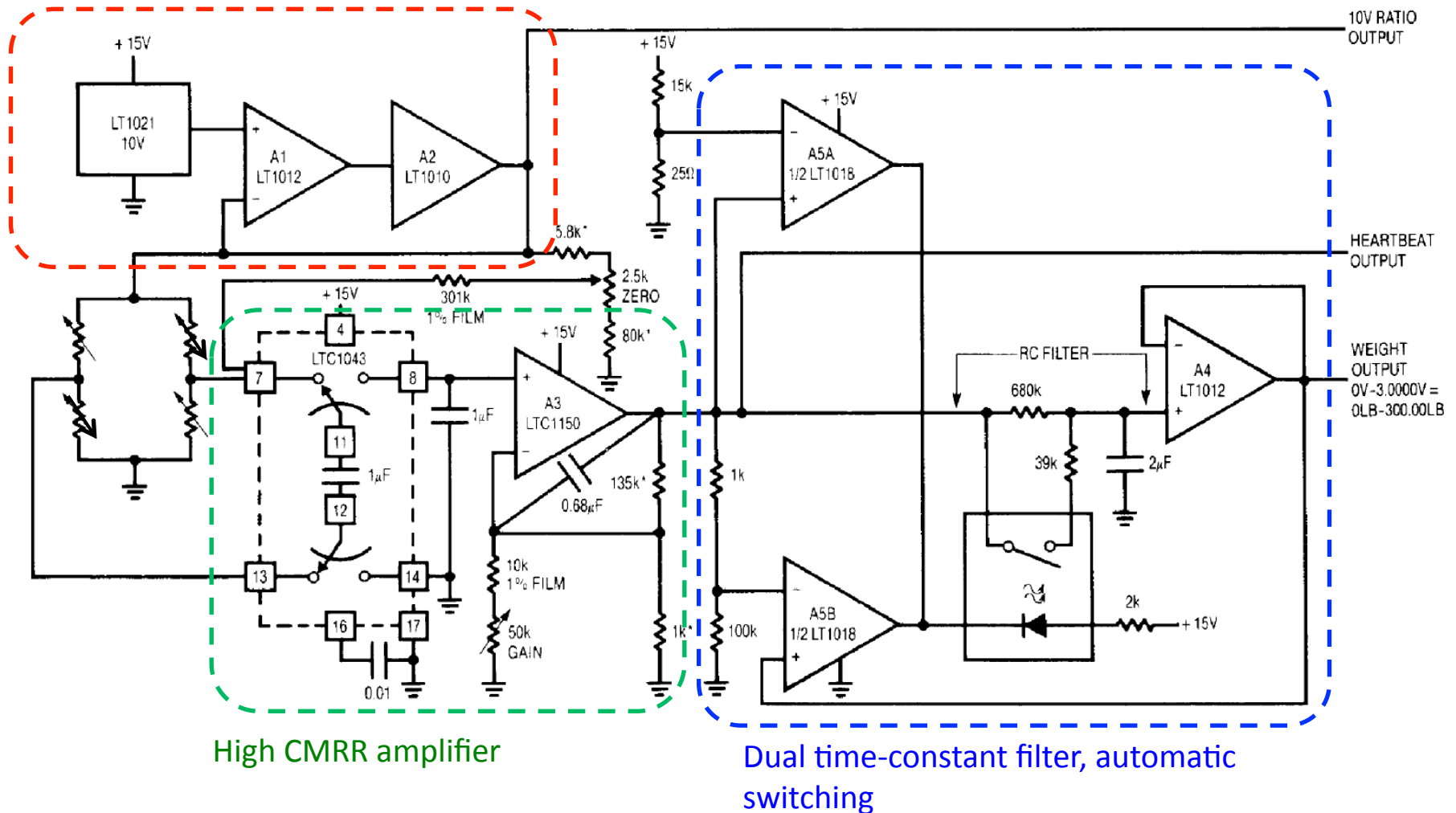


Neurocom Basic Balance Master®

Scale Signal Conditioning

Another bridge conditioning circuit...

Precision bridge excitation

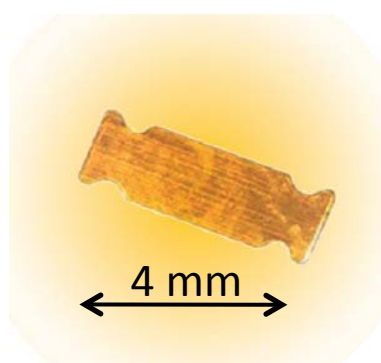


Accelerometers

- Accelerometers are based on Newton's law $F=ma$. They measure the force generated by a proof mass (or seismic mass), usually through the displacement of a mounting spring.
- Applications in activity monitoring, pacemakers, biomechanics (joint acceleration), CPR (depth and rate of compression)...
- Primary technologies for medical applications:
 - Piezoelectric
 - Capacitive
 - (Piezoresistive)
- MEMS technology enables very small devices.
Volume under one cubic millimeter

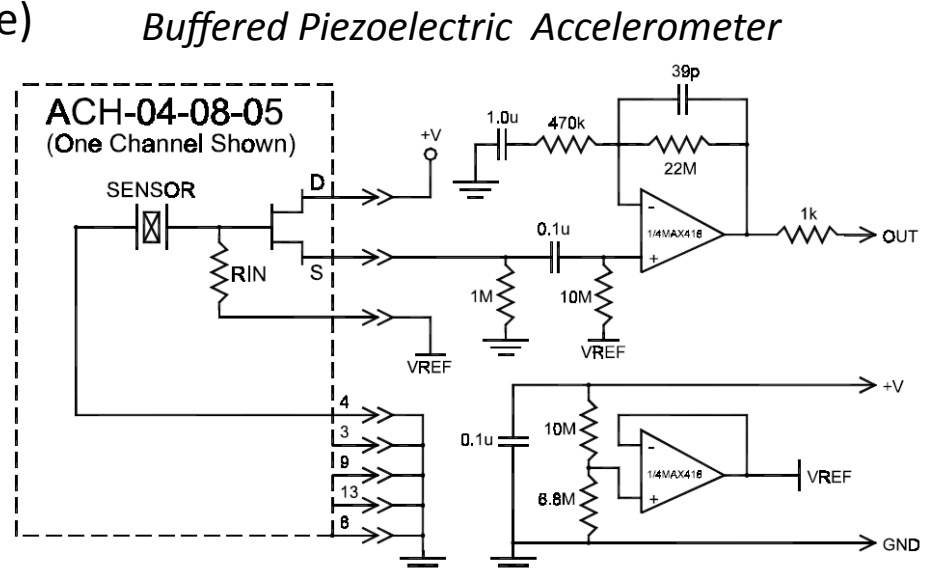
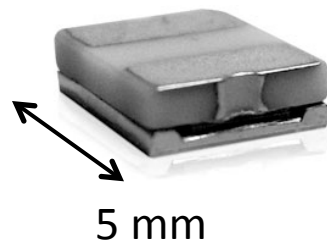
Piezoelectric Accelerometers

- Piezoelectric material compressed by proof mass generates charges.
- Do not require power.
Good reason for pacemaker use...
- High output impedance, unless buffered.
Require adequate conditioning (charge amplifier)
- No DC response (but can be close) *Buffered Piezoelectric Accelerometer*



*Piezo film used in pacemaker
(Measurement Specialties, used without permission)*

Endevco 12M1A



Source: Measurement Specialties ACH-04-08-05 datasheet

Reminder: Charge Amplifiers

Piezoelectric accelerometers (as other charge generating sensors) require high sensitivity charge amplifiers (integrators).

Circuit requires:

- Low I_b , high Z_{in} op-amps
- Low-leakage capacitor
- Large resistors

T network:

$$R_{eq} = R_1 \cdot \left(1 + \frac{R_2}{R_3}\right)$$

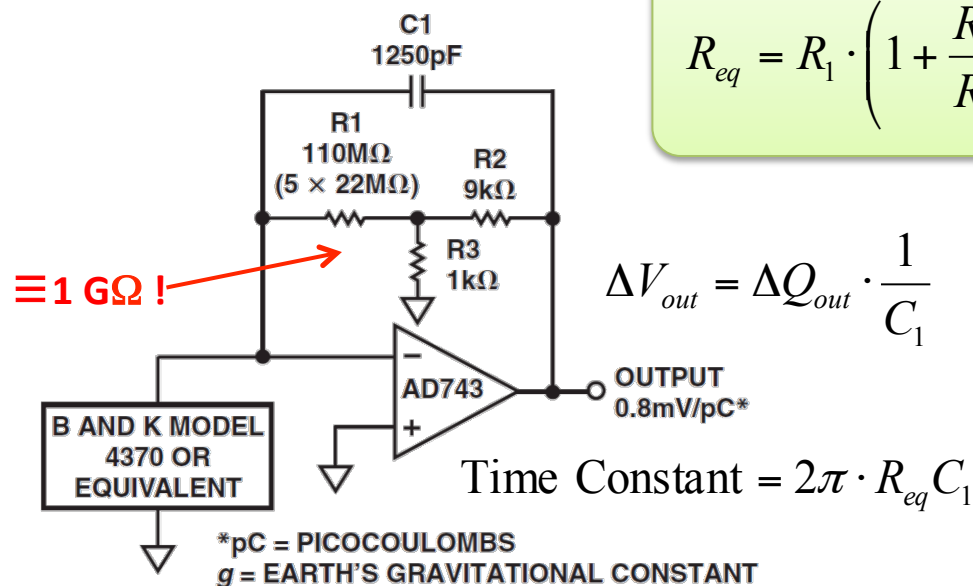


Figure 14a. Basic Accelerometer Circuit

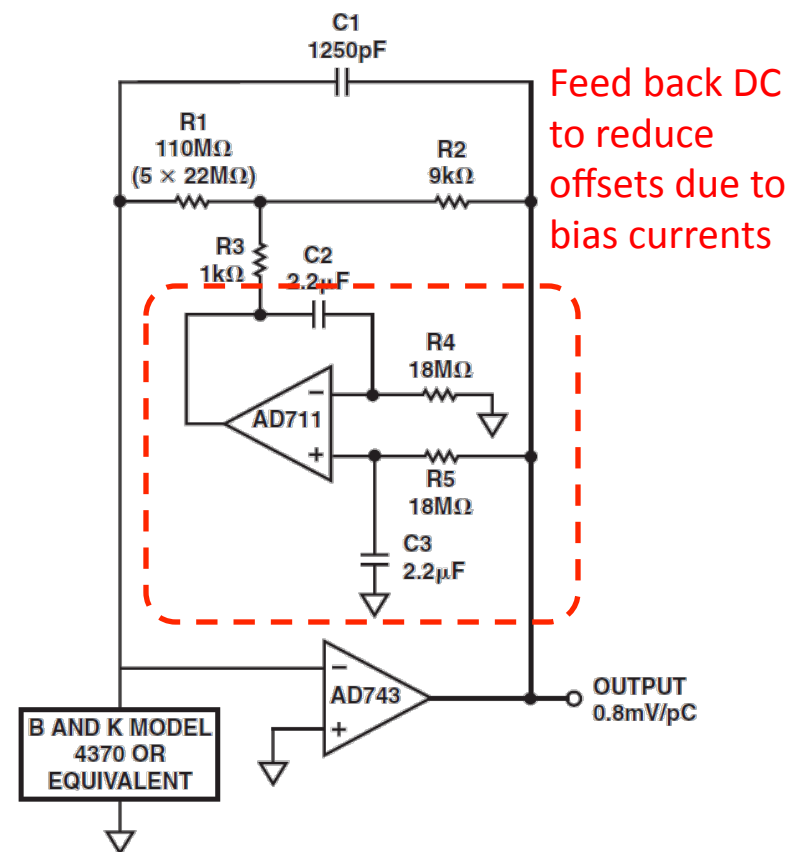
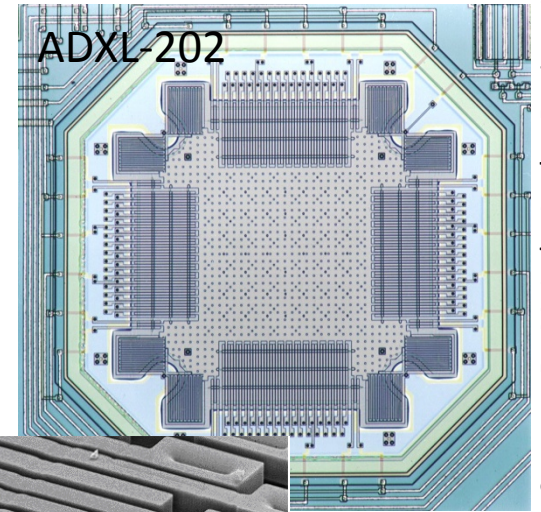


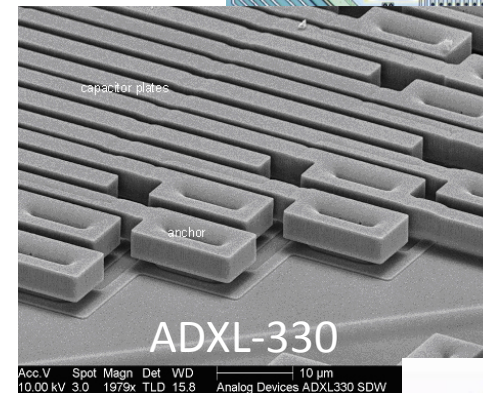
Figure 14b. Accelerometer Circuit Using a DC Servo Amplifier

Capacitive Accelerometers

- A proof mass is the movable plate of a variable capacitor. Changes in capacitance is related to movement of the proof mass against a spring OR force to hold proof mass in place [force-feedback].
- Conditioned output
Buffered, low-impedance, analog or digital
- Require power
3-axis usually requires 500 μ W
- DC response
Useful for orientation determination
- Miniature 3-axis available
Analog Devices ADXL330,
ST Microelectronics, Freescale...



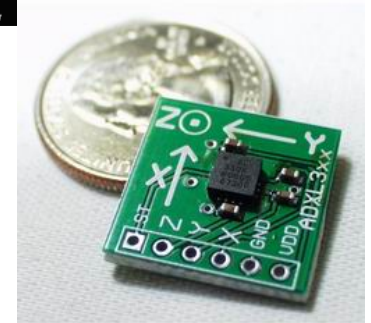
Courtesy Dr. M. Judy, Analog Devices, Inc.



Source: Chipworks,

ADXL-330

± 3 g, 4x4x1.5 mm
300 mV/g
380 μ g/Hz $^{1/2}$ noise
1.6 kHz BW (x,y)
320 μ A @ 3V



Application: CPR Quality Control

ZOLL® AED Plus Automated External Defibrillator

Measures compression depth
(double integration of acceleration)
and rate to guide CPR

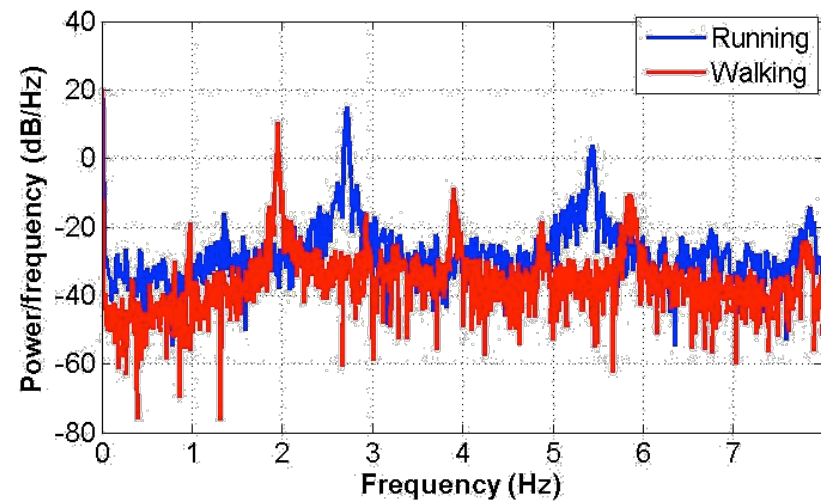
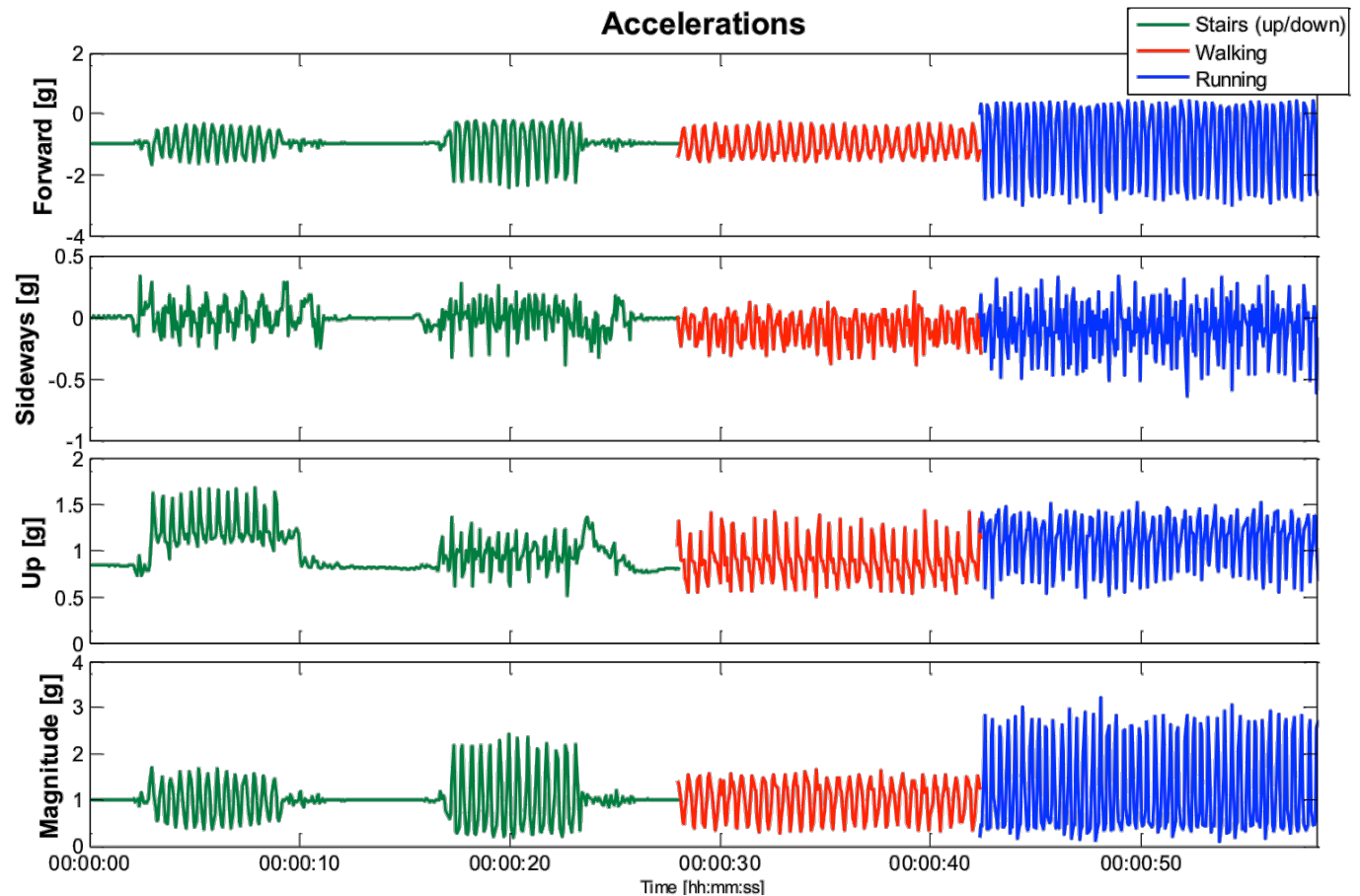


Accelerometer $\iint a \, dt \rightarrow$ Compression depth



Source: www.zoll.com, product brochure & video

Application: Activity Monitoring



Technology

Chemical Sensing

Chemical Sensing

- Measurements of metabolites, blood gases, electrolytes, molecules (hormones,...) and proteins in bodily fluids.
- Historically done in centralized laboratories, but now pushed to point-of-care.
- Large field, impossible to cover all here.
- Focus on a few electrochemical examples for blood chemistry analysis, representative of current point-of-care systems.

Blood Chemistry

Table 10.1 Critical-Care Analytes and Their Normal Ranges in Blood

Blood Gases and Related Parameters		Electrolytes		Metabolites	
P_{O_2}	80–104 mm Hg	Na^+	135–155 mmol/l	Glucose	70–110 mg/ 100 ml
P_{CO_2}	33–48 mm Hg	K^+	3.6–5.5 mmol/l	Lactate	3–7 mg/ 100 ml
pH	7.31–7.45	Ca^{2+}	1.14–1.31 mmol/l	Creatinine	0.9–1.4 mg/ 100 ml
Hematocrit	40–54%	Cl^-	98–109 mmol/l	Urea	8–26 mg/ 100 ml
Total hemoglobin	13–18 g/100 ml				
O_2 -saturation	95–100%				

SOURCE: M. E. Collison and M. E. Meyerhoff, “Chemical sensors for bedside monitoring of critically ill patients,” *Anal. Chem.*, 1990, 62, 425A–437A.

Electrochemical Sensors

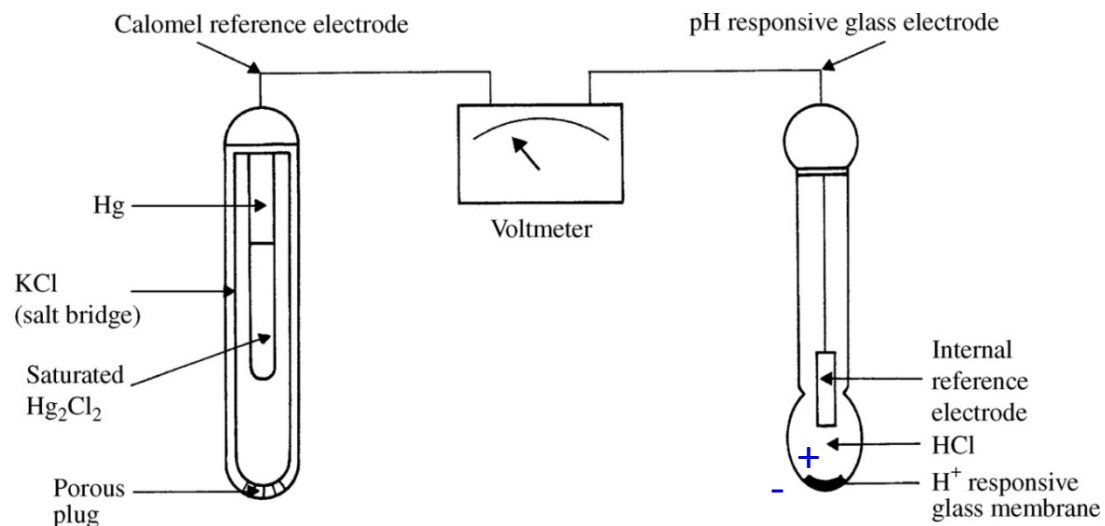
- Electrochemical sensors can be categorized in two groups:
 - *Potentiometric*
The sensor measures the potential of an electrochemical reaction (e.g., pH).
 - *Amperometric*
The sensor measures the current generated by an electrochemical reaction at a given potential (e.g., O₂).
- As examples, we will see one of each.
- Beside electrochemical sensors, other classes of chemical sensors include conductimetric sensors, or ISFET/ChemFET sensors.

pH - Potentiometric Sensor

- A ion-selective membrane separates the specimen from a cavity with a fixed concentration of H^+ (pH).
- A *potential* develops across the membrane, defined by the Nernst equation.

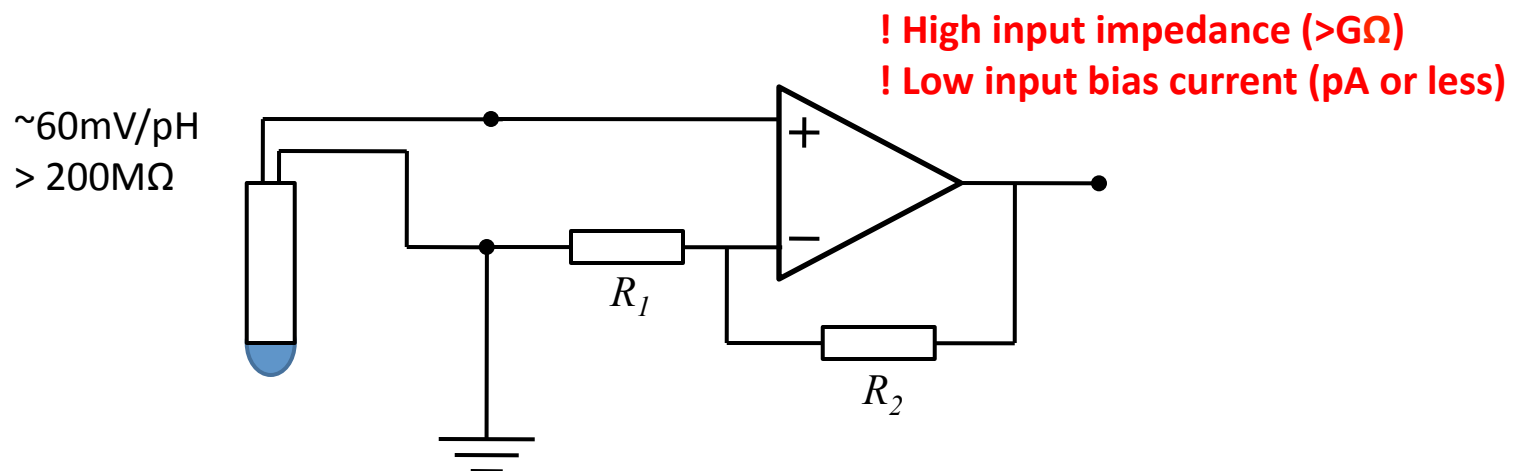
Recall the lecture on cell membrane potential!

- The measurement of this potential, with respect to a reference, yields the pH.



pH - Potentiometric Sensor

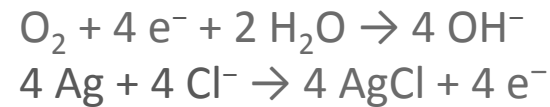
- The pH meter is essentially a two-electrode system.
- The reference electrode is often an Ag/AgCl electrode.
- Potentials generated are in the millivolt range.
~60mV/pH unit
- Output impedance of the pH electrode is very high (ion-sensitive membrane highly resistive).
- Signal conditioning straightforward.



O₂ – Amperometric Sensor

Dissolved oxygen concentration can be sensed by plateau current proportional to rate of reduction at a noble metal cathode (*Clark polarographic method*):

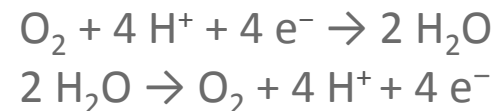
Cathode (reduction of O₂) :
Anode (oxidation of Ag):



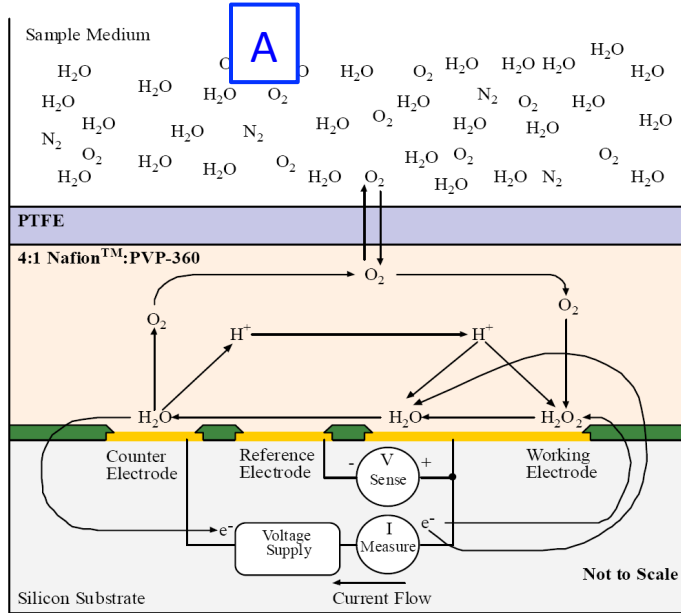
A

Or

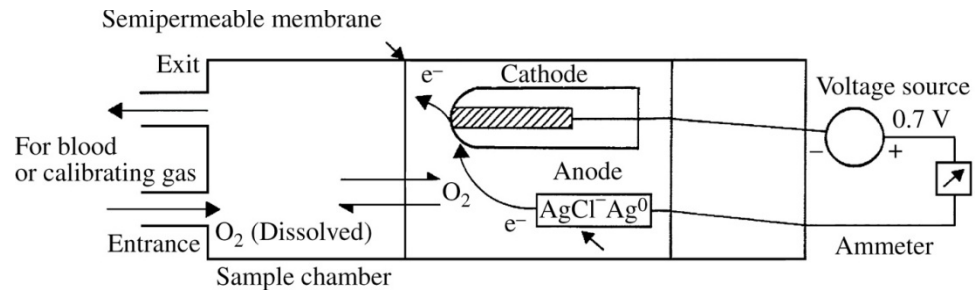
Cathode (reduction of O₂) :
Anode (oxidation of H₂O):



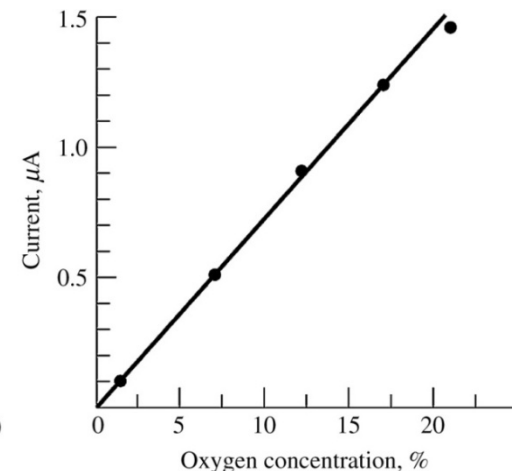
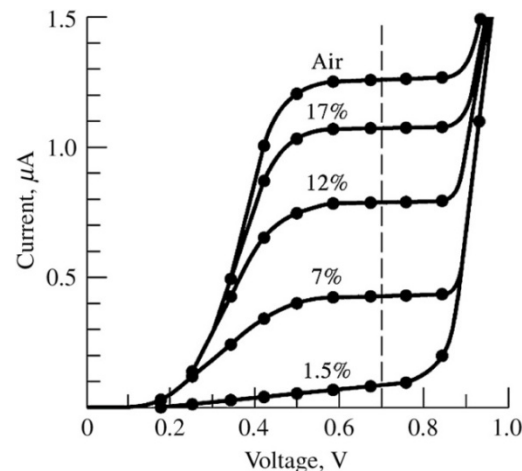
B



Source: G. McLaughlin, PhD thesis, Stanford University.



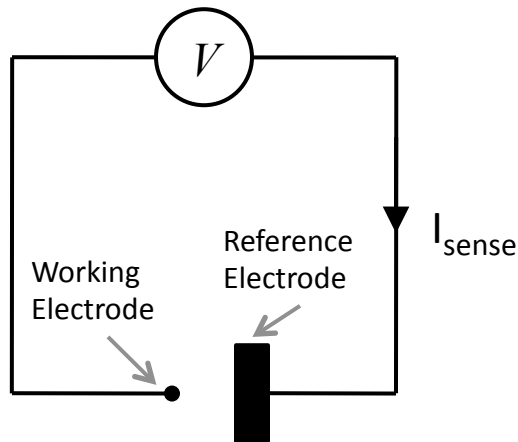
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O₂ Sensing - Potentiostat

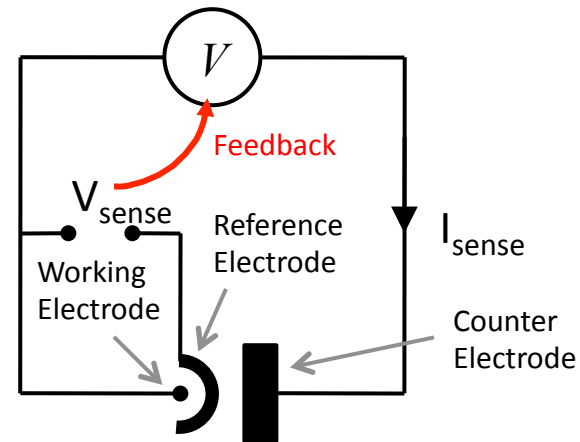
- A potentiostat is typically used to hold the cathode at a fixed voltage with respect to a reference (such as Ag/AgCl electrode), and measure the resulting current.
- Reference electrode must keep a stable potential, which is not trivial if current is passing (redox events).
- Solution: Decouple voltage sensing (reference) from current passing (counter). This is similar to a Kelvin connection.

Two-electrode configuration



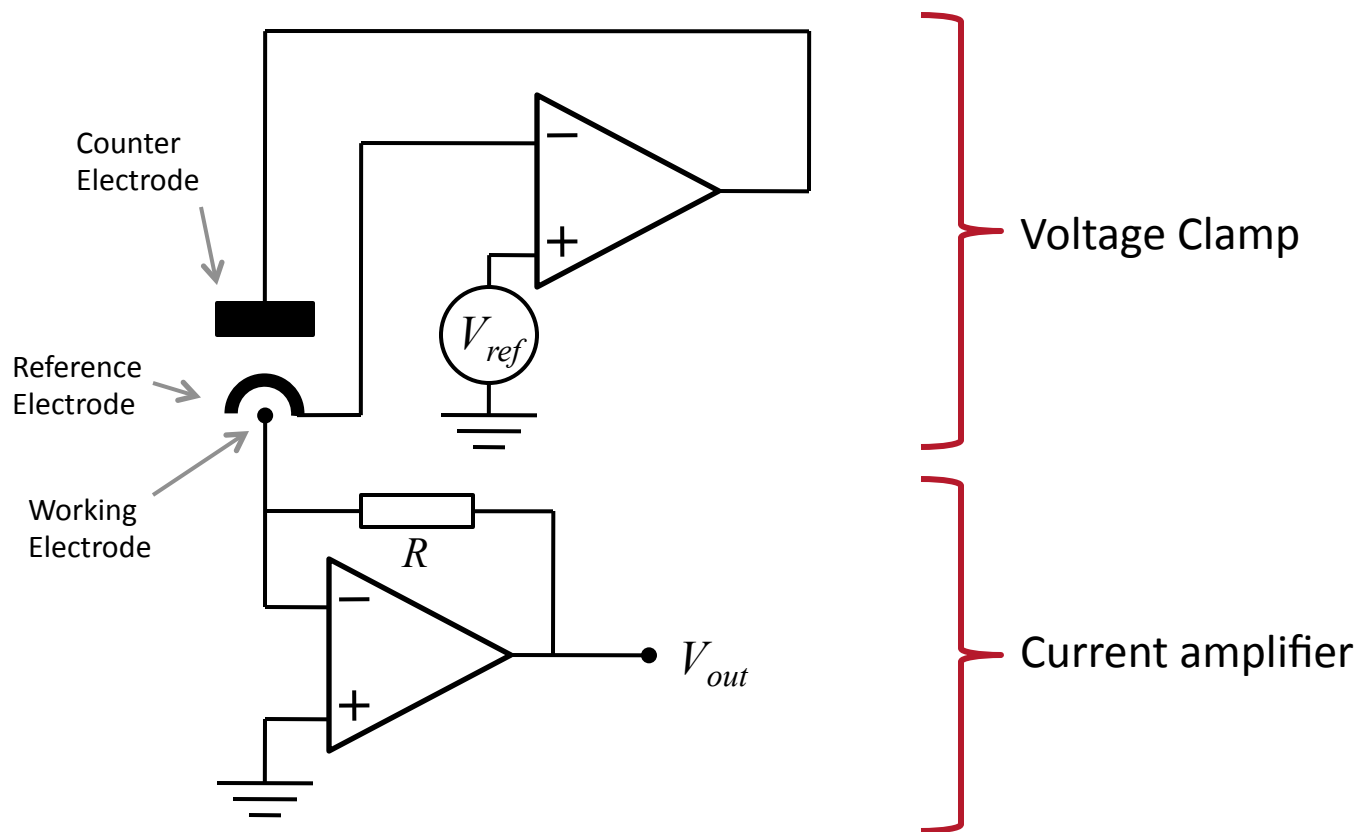
Suitable with stable reference/counter electrode (e.g., Ag/AgCl, Calomel)

Three-electrode configuration



Better with pseudo references (e.g., Au, Pt)

Basic Three-Electrode Potentiostat

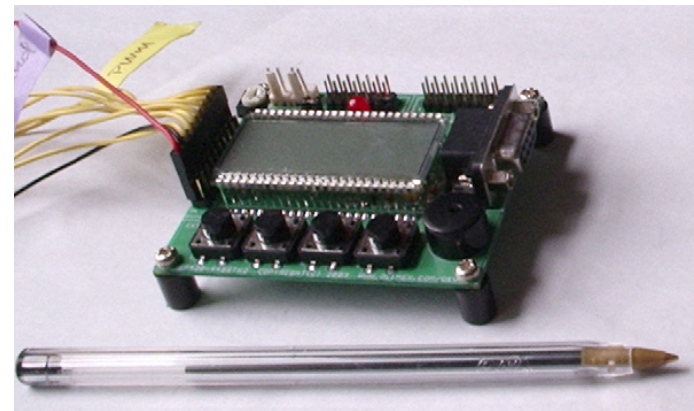
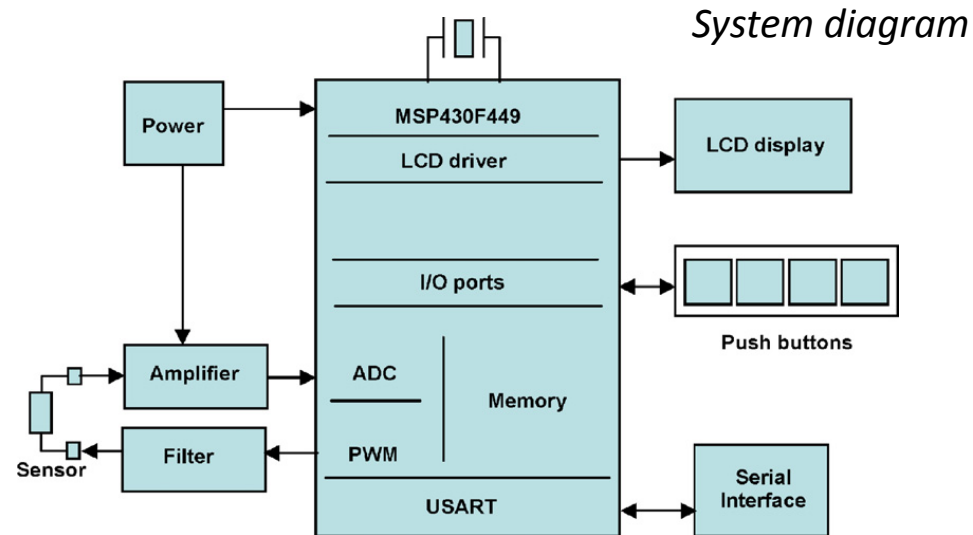
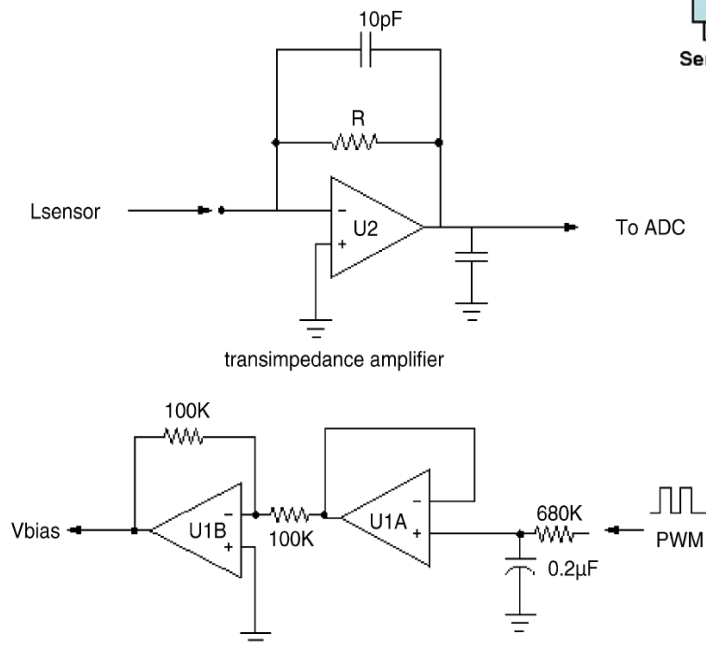


Both operational amplifiers should be electrometer-grade (input impedance > G Ω , bias current < fA)

Low-Power Electrochemical Unit

Low power, two-electrode potentiostat:

- Microcontroller-driven
- PWM-controlled reference voltage



Source: *An embedded system for portable electrochemical detection*, Kwakye & Baeumner, Sensors Actuators B, 2007

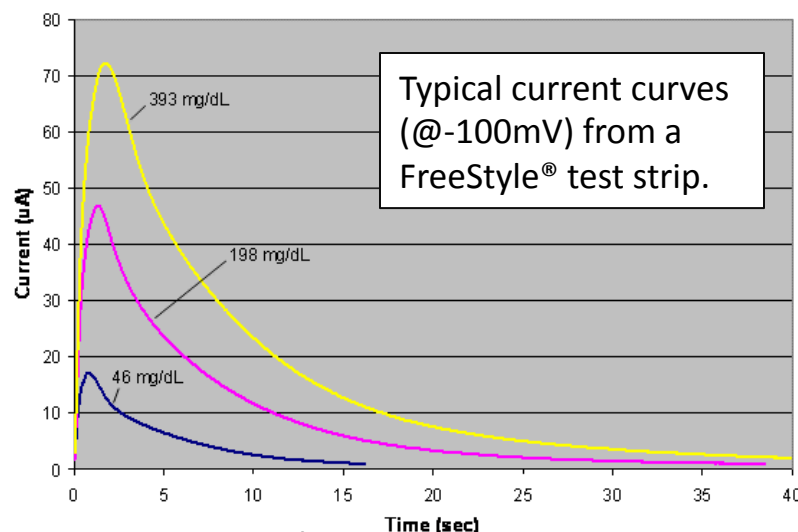
Diabetes and Glucose Sensing

- Diabetes is a condition where the body cannot process glucose into energy, due to lack of insulin production (Type I), or inadequate response to insulin (Type II, most common).
- ~8% of US population (25 million) has diabetes!
- Diabetes management requires frequent (~ four times a day) blood glucose level measurements to direct insulin intake/diet.
- $25 \text{ million} \times 4 \times 365 = \mathbf{36.5 \text{ billion tests}}$ per year!
- Cost of diabetes management (per person): \$1000-2000/year.
- Glucose sensing performed with handheld blood glucose sensors, using either optical or electrochemical (growing majority) sensing.

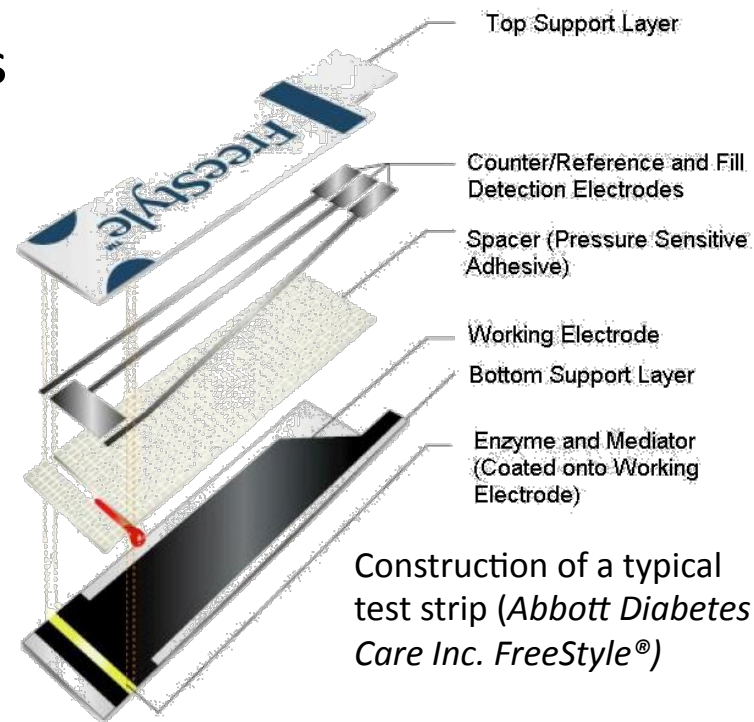
Electrochemical Glucose Sensors

Principle:

- Oxidation of glucose by an **enzyme** (e.g., glucose oxidase) generates either directly a measurable *current*, or another species (O_2 , H_2O_2) measurable by a conventional sensor.
- Most commonly, the measurement is *amperometric* – current is measured for an applied potential:

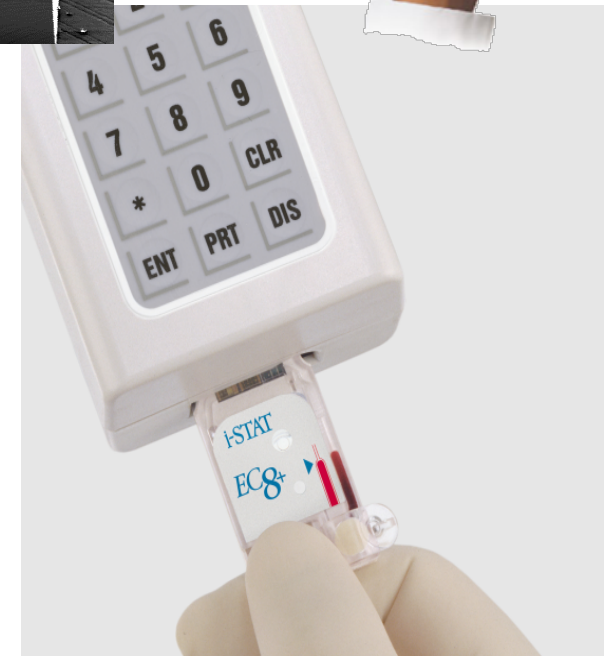
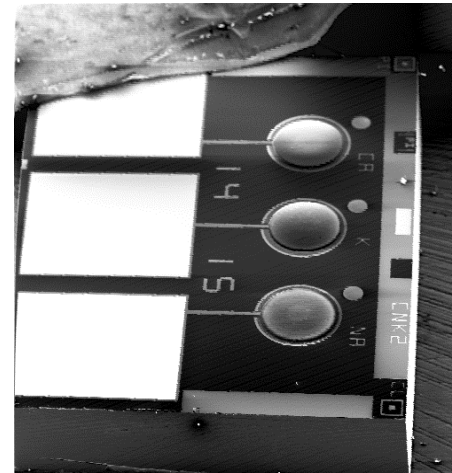
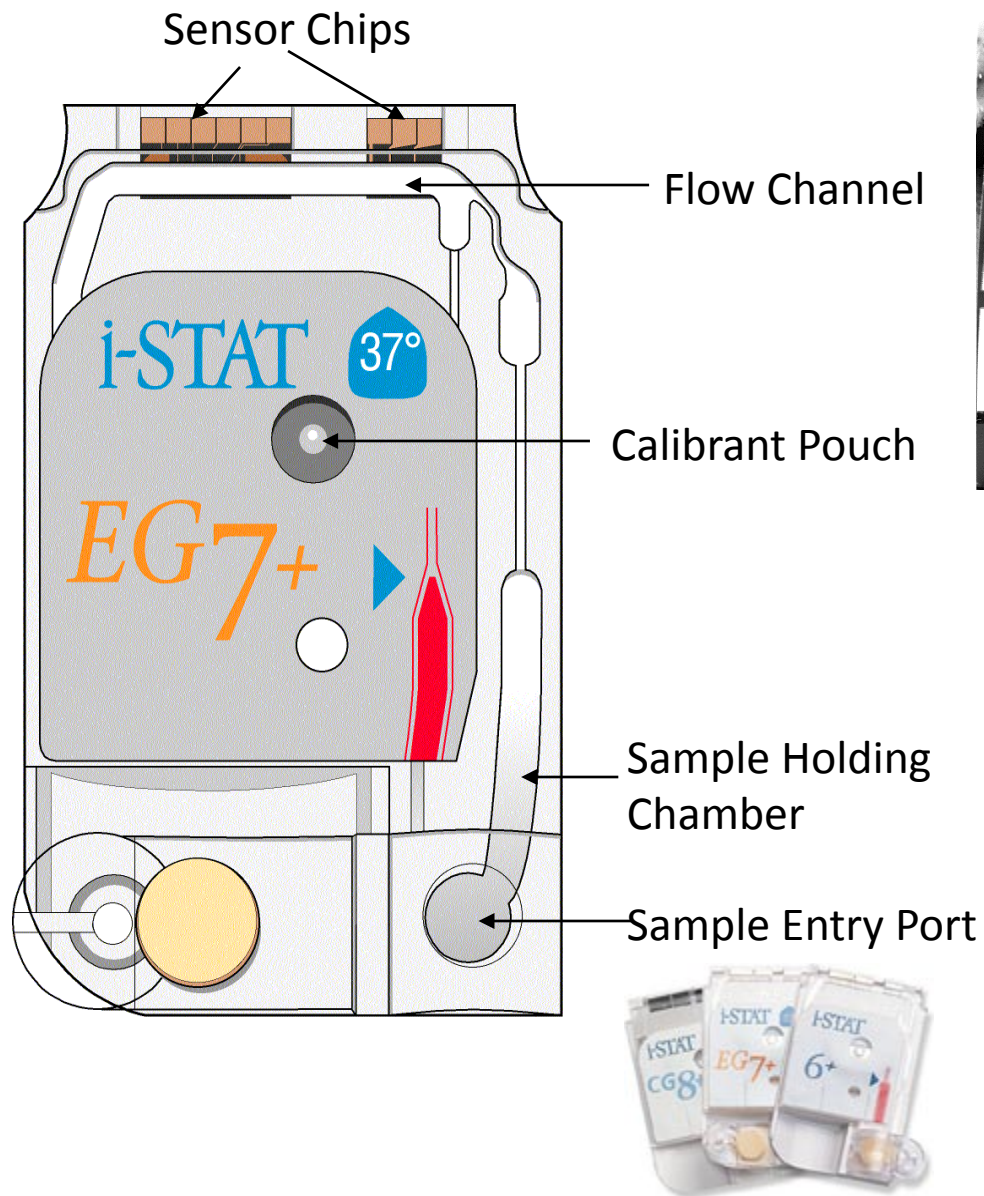


Example: OneTouch® blood glucose sensor from LifeScan.



Source: Feldman, *Electrochemistry Encyclopedia*, 2009 (<http://electrochem.cwru.edu/encycl/>)

Blood Analyzer Example – *i-STAT*®



Courtesy Dr. Anca Varlan, I-Stat. Slide from EE312

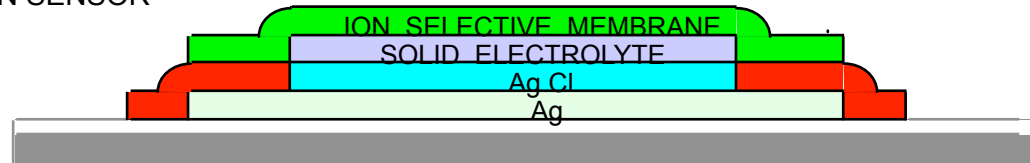
Blood Analyzer Example – *i-STAT*®

Sensor structure and chemistry for basic blood chemistry.

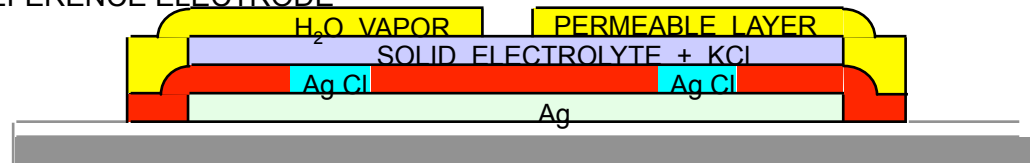
Based on similar principles, sensing has been extended to many other relevant blood analytes :

- Glucose
- Creatinine
- Cardiac markers (troponin, BNP, CK-MB)
- Urea nitrogen
- ...

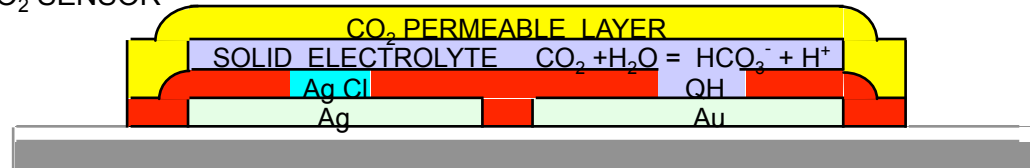
ION SENSOR



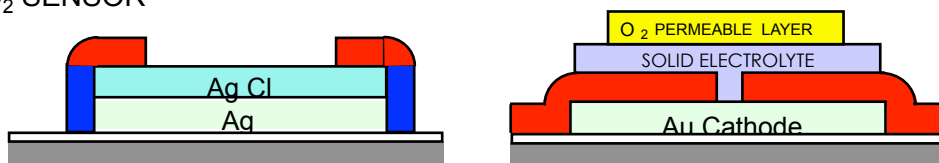
REFERENCE ELECTRODE



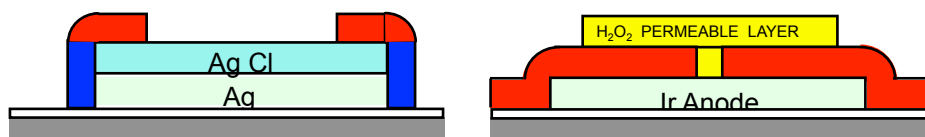
CO₂ SENSOR



O₂ SENSOR



H₂O₂ SENSOR



Lab this week

Project Time !!!